

## A Novel Metabolite of Tinidazole Involving Nitro-group Migration

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### Introduction

Tinidazole (1) is one of a group of nitroimidazoles which are effective antiprotozoal agents (Nord, 1982). A major urinary metabolite of (1) is a hitherto unidentified bright-yellow compound 'metabolite 3' which accounts for 14% (Wood, Rycroft and Monro, 1973) and 20% (unpublished data) of the dose of tinidazole in dog and man respectively. Recently we have isolated sufficient metabolite 3 to permit unequivocal identification of its structure.

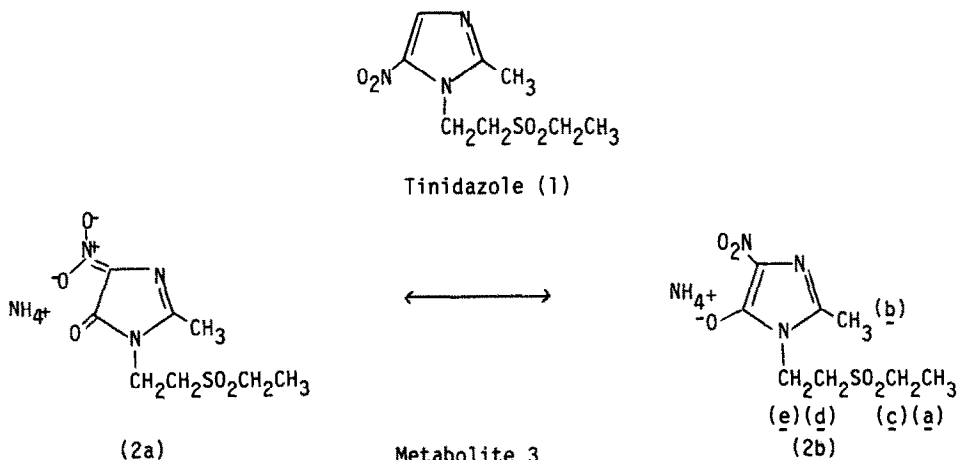
### Experimental

After oral administration of  $^{14}\text{C}$ -tinidazole to dogs, metabolite 3 was isolated from the urine using two reversed phase high-performance liquid chromatography systems. After an initial 'clean-up', metabolite 3 was separated from the glucuronic acid conjugate of hydroxymethyltinidazole using a semi-preparative column prepacked with Spherisorb 10 O.D.S. and eluted with 0.1M ammonium acetate-methanol (95 : 5 v/v). Metabolite 3 was finally isolated by slow crystallisation from rigorously dried methanol at 10°C.

### Results

The ultra-violet spectrum ( $\lambda_{\text{max}}$  370 nm,  $\epsilon$  17000) and  $^1\text{H}$ -nmr spectrum in  $\text{MeOH-d}_4$  [ $\delta$  1.33 (t,  $J=7.5$  Hz) 3H(a);  $\delta$  2.30 (s) 3H (b);  $\delta$  3.09 (q,  $J=7.5$  Hz) 2H (c);  $\delta$  3.51 (t,  $J=6.3$  Hz) 2H (d);  $\delta$  4.13 (t,  $J=6.3$  Hz) 2H (e)] of metabolite 3 indicated a ring-hydroxylated tinidazole structure. This was supported by the FAB mass spectrum ( $m/e$  262) and the elemental analysis (Found (mean) C, 34.6; H, 5.9; N, 20.3; S, 11.8%. Calculated for  $\text{C}_8\text{H}_{12}\text{N}_3\text{O}_5\text{S} \cdot \text{NH}_4$ : C, 34.3; H, 5.7; N, 20.0; S, 11.4%) which suggested that metabolite 3 had been isolated as the ammonium salt of ring-hydroxylated tinidazole.

The structure of metabolite 3 was shown by X-ray diffraction analysis to be the ammonium salt of the 4-nitroimidazolin-5-one (2a). In the light of this unexpected result, tinidazole itself was also subjected to X-ray crystallographic analysis and its structure confirmed as the expected 5-nitro-compound (1).



### Discussion

Formulation of the nitro-group in metabolite 3 as  $C=NO_2^-$  is supported by comparison of the bond lengths for metabolite 3 with those in a range of nitromidazole compounds, including tinidazole, and with those in a compound incorporating both types of nitro-group (Messmer and Palenik, 1969).

However, it is possible that in solution the compound exists as a resonance hybrid of forms (2a) and (2b) has been suggested for ring-hydroxylated nitrofurantoin (Olivard et al., 1976) and a hydroxylated metabolite of ipronidazole (Weiss et al., 1981). The mechanism of formation of this unusual metabolite presumably involves a rearrangement perhaps analogous to the "NIH shift" (Jerina and Daly, 1974 in which intramolecular migration of the 5-nitro-group to the 4-position is preceded by epoxidation of the 4, 5-double bond.

### References

- D. M. Jerina and J. M. Daly (1974) Arene oxides: a new aspect of drug metabolism. *Science*, 185, 573
- G. G. Messmer and G. J. Palenik (1969) The crystal structure of a meisenheimer complex: the potassium methoxide adduct of 4-methoxy-5, 7-dinitrobenzofurazan. *J. Chem. Soc., Chem. Commun.*, 470
- C. E. Nord (1982) Microbiological properties of tinidazole: spectrum, activity and ecological considerations. *J. Antimicrob. Chemother.* 10, Suppl. A. 35
- J. Olivard, G. M. Rose, G. M. Klein and J. P. Heotis (1976) Metabolic and photochemical hydroxylation of 5-nitro-2-furancarboxaldehyde derivatives. *J. Med. Chem.* 19, 729
- G. Weiss, N. Rose, P. Duke and T. H. Williams (1981) A major rat faecal metabolite of ipronidazole. *Xenobiotica* 11, 207
- B. A. Wood, J. K. Faulkner and A. M. Monro (1982) The Pharmacokinetics, metabolism and tissue distribution of tinidazole. *J. Antimicrob. Chemother.* 10, Suppl. A, 43
- B. A. Wood, D. Rycroft and A. M. Monro (1973) The metabolism of tinidazole in the rat and dog. *Xenobiotica* 3, 801